PRELIMINARY NOTES

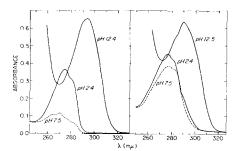
вва 61166

The effect of a competitive inhibitor on the acetylation of tyrosyl and lysyl residues of staphylococcal nuclease

About five of the seven tyrosine residues of staphylococcal nuclease appear to be accessible to solvent, as judged by studies of ultraviolet absorption¹. The binding of deoxythymidine 3',5'-diphosphate (pdTp) causes a large red shift in tyrosine absorption¹, displacement of the tyrosine titration curve², decreased tyrosylexposure to neutral solvents¹ and quenching of tyrosyl fluorescence³.

This report describes the reactivity of the tyrosyl residues of nuclease toward chemical modification by acetylimidazole, a reagent reported to be relatively specific for tyrosine acetylation^{4–6}. Five tyrosyl and 9–10 lysyl residues are acetylated in the native enzyme with almost complete loss of deoxyribonuclease and ribonuclease activities; there is complete return of deoxyribonuclease and partial return of ribonuclease activities on O-deacetylation. Only 2 tyrosyl and 8–9 lysyl residues are acetylated in the presence of pdTp and Ca²⁺ and there is no loss of deoxyribonuclease activity; a small fall in ribonuclease activity occurs which is reversed on O-deacetylation.

Acetylation is performed by adding aliquots of 0.5 M acetylimidazole freshly dissolved in 10 mM Tris buffer (pH 7.5) to nuclease (2 mg/ml) dissolved in the same



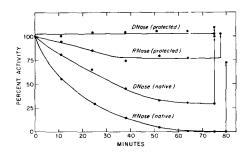


Fig. 1. Ultraviolet absorption spectra used to estimate the content of O-acetyl tyrosine. Right: A 1-ml sample of nuclease, acetylated with a 65-fold molar excess of acetylimidazole, in 5 mM Tris buffer (pH 7.5) was O-deacetylated by addition of 50 μ l of 5 M NaOH (pH 12.5) and then acidified with 50 μ l of 6 M HCl (pH 2.4). The protein concentration (2.9·10⁻⁵ M) was obtained from the spectrum in acid, which is identical to that of native nuclease. Left: 0.27 mM N-O-diacetyl tyrosine subjected to the same procedures as nuclease.

Fig. 2. Effect of acetylation with 50-fold molar excess of acetylimidazole on deoxyribonuclease (DNase) and ribonuclease (RNase) activities in the presence and absence of pdTp and CaCl₂. Arrows (at right) show change in activity occurring after alkaline deacetylation.

buffer. After 1 h (24°) the mixture is passed through Sephadex G-25 with 5 mM Tris buffer (pH 7.5). The O-acetyl tyrosine content is determined as described in Fig. 1. The $\Delta \epsilon_M$ at 275 m μ for O-deacetylation of N-O-diacetyl tyrosine was 1150.

Five tyrosyl residues are readily acetylated in the native enzyme with con-

418 PRELIMINARY NOTES

TABLE I EFFECT OF NUCLEOTIDES AND Ca^{2+} on the acetylation of tyrosine residues of nuclease Nuclease (0.10 mM) was reacted with a 70-fold excess of acetylimidazole. Ca^{2+} when present was 10 mM.

Addition	Nucleotide concn. (mM)	Deoxyriboni (%) ⁷	Tyrosines acetylated	
		Acetylated	Deacetylated	
None		20	96	5.5
pdTp	0.70	40	100	4.2
Ca2+		18	97	5·7
$pdTp + Ca^{2+}$	0.19	115	110	2.4
$pdTp + Ca^{2+}$	0.70	111	113	2.1
Urea (8 M)	,	6	80	7.3
$pdTp + Ca^{2+} + urea (8 M)$	0.70	10	81	7.1
$\widehat{\mathrm{CMP}} + \mathrm{Ca^{2+}}$	0.36	29	75	5.6

comitant loss of 80% of the deoxyribonuclease activity (Table I). If Ca^{2+} is present, pdTp prevents this loss of activity and protects 3 tyrosines from reacting. In 8 M urea all 7 tyrosyls are acetylated, and there is no protection by the nucleotide. A non-binding (CMP) nucleotide has no effect. Removal of the O-acetyl groups by alkaline treatment restores to normal the deoxyribonuclease activity (Table I), indicating that tyrosyl residues may be essential for catalytic integrity.

The fall in ribonuclease activity is more rapid and extensive than the fall in deoxyribonuclease activity (Fig. 2). Ribonuclease activity in the presence of pdTp falls by 25% whereas deoxyribonuclease activity is unchanged. Since this activity is restored on O-deacetylation, it may be that one or both of the two tyrosyls acetylated is involved in RNA hydrolysis. The ribonuclease activity of the enzyme, acetylated without nucleotide, is only partially restored on deacetylation, suggesting that a residue forming an alkaline-stable derivative is involved in RNA but not DNA hydrolysis.

TABLE II

ACETYL CONTENT OF NUCLEASE TREATED WITH ACETYLIMIDAZOLE

pdTp (0.60 mM) and CaCl, (10 mM) were added to the solution of native enzyme (0.17 mM).

Enzyme	Molar excess of reagent	Moles/mole enzyme							
		Before alkaline treatment			After alkaline treatment				
		Acetyl tyrosine	[14C]- Acetyl groups	% Deoxy- ribo- nuclease activity	Acetyl tyrosine	[14C]- Acetyl groups	Lysine content*	% Deoxy- ribo- nuclease activity	
Native	10	0.9	2.5	88	О	1.6		100	
Protected	10	0.2	1.5	100	0	I.I		100	
Native	50	3.0	9.8	48	o	7.3		100	
Native	100	5.2	14.4	7	О	9.6	10.2	94	
Protected	100	2.3	10.1	94	o	7.9	9.0	100	

^{*} Determined by deamination with nitrous acid, followed by acid hydrolysis and amino acid analysis.

Biochim. Biophys. Acta, 159 (1968) 417-419

PRELIMINARY NOTES 419

Studies using [14C]acetylimidazole reveal that the number of alkaline-labile acetyl groups is equal to the number of spectrally determined O-acetyl tyrosines (Table II). This excludes aliphatic hydroxyl acetylation. However, large amounts of alkaline-stable [14C] acetyl groups are present. Acetyl histidine is probably not present since no loss of radioactivity occurs after 4 h at pH 2.o. It is likely that lysine accounts for the large content of [14C] acetyl as suggested by studies of nitrous acid deamination (Table II).

The inhibitor, pdTp, appears to protect at least one lysine residue from acetylation (Table II). Further support for this is obtained from peptide maps of trypsin digests of nuclease treated with 10-, 20-, and 100-fold excess of [14C] acetylimidazole, which show one much less intense radioactive peptide in the maps of samples a etylated in the presence of pdTp. Amino acid analysis, after elution of the peptide, suggests that the lysine residue at position 84 (ref. 10) is protected from acetylation.

It is of interest that nuclease has full deoxyribonuclease activity in spite of having 8-10 of its amino groups blocked. Moreover, such a modified nuclease does not differ significantly from the native enzyme in many catalytic properties, i.e., pH optimum, optimum Ca^{2+} concentration, K_m and v_{max} for DNA, and inhibition by pdTp.

Laboratory of Chemical Biology, National Institute of Arthritis and Metabolic Diseases, Bethesda, Md. 20014 (U.S.A.)

Pedro Cuatrecasas SARA FUCHS* CHRISTIAN B. ANFINSEN

```
I P. CUATRECASAS, S. FUCHS AND C. B. ANFINSEN, J. Biol. Chem., 242 (1967) 4759.
```

Received January 22nd, 1968

Biochim. Biophys. Acta, 159 (1968) 417-419

² P. Cuatrecasas, S. Fuchs and C. B. Anfinsen, J. Biol. Chem., 242 (1967) 3063.

³ P. Cuatrecasas, H. Edelhoch and C. B. Anfinsen, Proc. Natl. Acad. Sci. U.S., 58 (1967) 2043.

⁴ J. F. RIORDAN, W. E. C. WACKER AND B. L. VALLEE, Biochemistry, 4 (1965) 1758.

 ⁵ G. E. PERLMANN, J. Biol. Chem., 241 (1966) 153.
 6 E. L. Pugh and B. L. Horecker, Biochem. Biophys. Res. Commun., 26 (1967) 36c.

⁷ P. CUATRECASAS, S. FUCHS AND C. B. ANFINSEN, J. Biol. Chem., 242 (1967) 1541.

⁸ G. A. Bray, Anal. Biochem., 1 (1960) 279.

⁹ S. Fuchs, P. Cuatrecasas and C. B. Anfinsen, J. Biol. Chem., 242 (1967) 4768.

¹⁰ H. TANIUCHI, C. B. ANFINSEN AND A. SODJA, J. Biol. Chem., 242 (1967) 4752.

^{*} Visiting Scientist ,on leave from the Weizmann Institute of Science, Rehovoth, Israel.